

<b>Atomic Number Practice</b>						
<b>Q #</b>	<b>Element Name</b> <small>(don't forget mass # at end!)</small>	<b>Atomic Number</b>	<b># of Protons</b>	<b># of Neutrons</b>	<b># of Electrons</b>	<b>Mass Number</b>
1	Carbon-					12
2		8		8		
3	Hydrogen-					1
4			6			14
5	Hydrogen-			2		
6	Nitrogen-					14
7				1		2
8		92		146		
9	Cesium-			82		
10		11		12		
11			47			108
12	Tungsten-			110		
13				45		80
14			24			52
15				89		152
16	Silver-					107
17		76		114		

**Answer on Notebook Paper:**

1) How are the atomic number and the number of protons related to each other?	2) How do the number of protons, number of neutrons, and the mass number relate?	3) What is the one thing that determines the identity (name) of an atom?
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