Specific Heat of Brass Lab Calculations

|  | Trial \#1 | Trial \#2 | Trial \#3 | Average |
| :---: | :---: | :---: | :---: | :---: |
| $\Delta \mathrm{T}_{\text {water, }}\left(\mathrm{T}_{2}-\mathrm{T}_{1}\right)$ |  |  |  |  |
| $\Delta \mathrm{T}_{\text {brass }}\left(\mathrm{T}_{2}-\mathbf{1 0 0 ^ { \circ }} \mathbf{C}\right)$ |  |  |  |  |

Specific Heat of Brass Lab Calculations

|  | Trial \#1 | Trial \#2 | Trial \#3 | Average |
| :---: | :---: | :---: | :---: | :---: |
| $\Delta \mathrm{T}_{\text {water, }}\left(\mathrm{T}_{2}-\mathrm{T}_{1}\right)$ |  |  |  |  |
| $\Delta \mathrm{T}_{\text {brass }}\left(\mathrm{T}_{2}-\mathbf{- 1 0 0 ^ { \circ }} \mathbf{C}\right)$ |  |  |  |  |


|  | Solving for Qwater <br> $\mathbf{Q}_{\text {water }}=C_{\text {water }} \times \mathbf{m}_{\text {water }} \times \Delta T_{\text {water }}$ |
| :--- | :---: |
| $m_{\text {water }}=$ |  |
| $\Delta T_{\text {water }}=$ |  |
|  |  |


| $Q_{\text {brass }}=-Q_{\text {water }}$ |
| :---: |
|  |
|  |

$$
\mathrm{Q}_{\text {brass }}=\mathrm{C}_{\text {brass }} \times \mathrm{m}_{\text {brass }} \times \Delta \mathrm{T}_{\text {brass }}
$$

## Calculating \% error <br> $\%$ error $=\left|\frac{\text { your number }- \text { accepted number }}{\text { accepted number }}\right|$

## Calculating \% error

$\%$ error $=$ your number - accepted number
accepted number

