

Pure Substances
vs. Mixtures
Study Guide

Mrs. V Science
2014



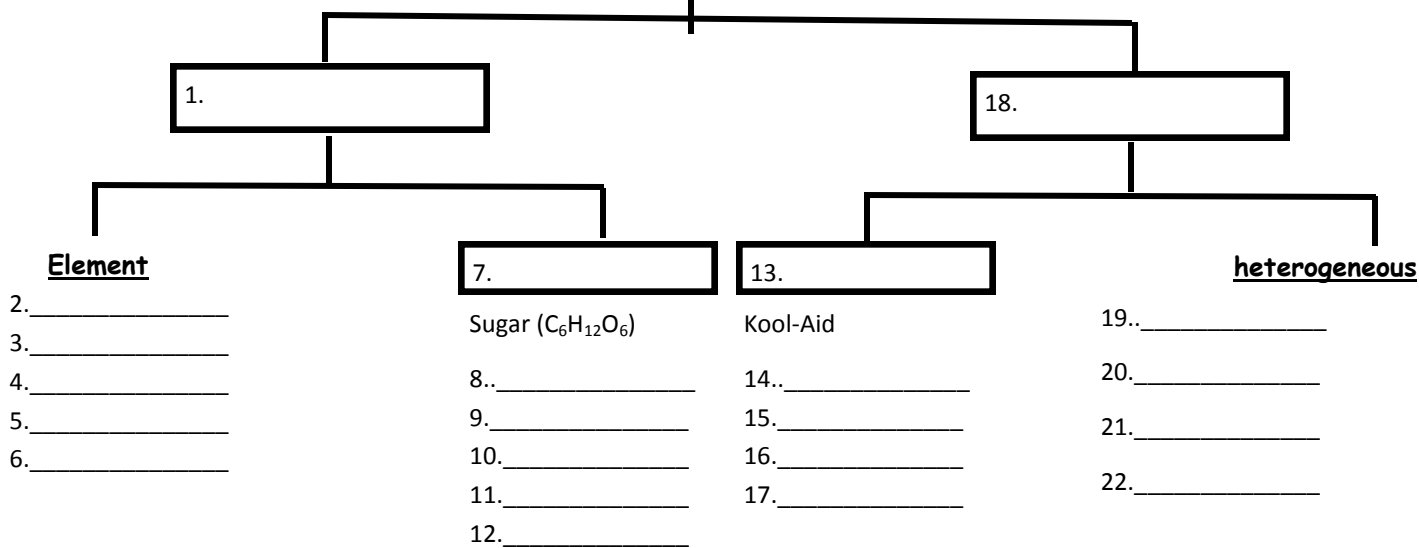
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Name: _____ Class: _____ Date: _____

Use the words and phrases to complete the tree map below. Some words/phrases may be used MORE than once.				
Pure Substances	Compound	Salt (NaCl)	Pizza	Salad
Mixtures	Sodium (Na)	Helium (He)	Air	chemically combined
not chemically combined	one kind of particle	Chlorine (Cl)	Salt Water	can see individual parts
Uniform appearance	Water (H ₂ O)	Molecule	Atom	homogeneous

Classifying Matter



Using CIRCLES to represent an atom of an element, and draw an example of each type of pure substance or mixture in the boxes below.

Element	Compound	Mixture of Elements	Mixture of Compounds	Mixture of Elements and Compounds

MATCHING:

- | | |
|--|-------------|
| 1. _____ The smallest unit of a compound that has the same properties as the compound. | A. Element |
| 2. _____ The smallest unit of an element that has the same properties of the element. | B. Molecule |
| 3. _____ A pure substance that can not be broken down into any simpler substances. | C. Atom |
| 4. _____ A pure substance made of two elements that are bonded together. | D. Compound |

Fill in the blank: Use the following words to complete each sentence.

Solid

Liquid

Gas

Plasma

1. A _____ does NOT have a definite shape or a definite volume.
2. A _____ has particles that are far apart and move very quickly.
3. A _____ has definite shape and definite volume.
4. A _____ is an electrically charged (ionized) gas.
5. A _____ has particles that slide past each other.
6. A _____ has a definite volume, but an Indefinite shape.
7. A _____ has particles that are tightly packed and vibrate in place.
8. Examples of a _____ are stars, lightning, & neon lights.
9. Examples of a _____ could be the helium in balloons.
10. Examples of a _____ include milk, honey, and juice.

A. Element

C. Mixture of Elements

E. Mixture of Elements and Compounds

B. Compound

D. Mixture of Compounds

