

Converting grams → moles → particles

Prediction: (tell me which beaker/substance you think will have the most number of molecules and **why**)

Summary:

Baking Soda

Formula:

Molar Mass:

Mass of full beaker:

Mass of empty beaker:

Mass of just the substance:

$g \rightarrow mol$

$mol \rightarrow particles$

$g \rightarrow mol \rightarrow particles$

Table Sugar

Formula:

Molar Mass:

Mass of full beaker:

Mass of empty beaker:

Mass of just the substance:

$g \rightarrow mol$

$mol \rightarrow particles$

$g \rightarrow mol \rightarrow particles$

Water

Formula:

Molar Mass:

Mass of full beaker:

Mass of empty beaker:

Mass of just the substance:

$g \rightarrow mol$

$mol \rightarrow particles$

$g \rightarrow mol \rightarrow particles$