|  |  |
| --- | --- |
| **#1** | **N2O4 (g)  <--------> 2NO2(g) ΔH = + 92 KJ** |
| **The Stress** | **[N2O4]** | **[NO2]** | **Right or Left** | **Reactants or Products** |
| [N2O4] is increased |  |  |  |  |
| [NO2] is increased |  |  |  |  |
| Temp is increased |  |  |  |  |
| [N2O4] is decreased |  |  |  |  |
| [H2] is decreased |  |  |  |  |
| [NO2] is decreased |  |  |  |  |
| Temp is decreased |  |  |  |  |
| **#2** | **4HCl (g)  + O2 (g) <--------> 2H2O(g) + 2Cl2 (g) + 98 KJ** |
| **The Stress** | **[O2]** | **[H2O]**  | **[HCl]** | **Right or Left** | **Reactants or Products** |
| [HCl] is increased |  |  |  |  |  |
| [H2O] is increased |  |  |  |  |  |
| [O2] is increased |  |  |  |  |  |
| Temp is increased |  |  |  |  |  |
| **#3** | **CaCO3 (s) + 170 KJ <----------> CaO (s) + CO2 (g)**Note: Adding solids or liquids and removing solids or liquids does not shift the equilibrium. This is because you cannot change the concentration of a pure liquid or solid as they are 100% pure. It is only a concentration change that will change the # of collisions and hence shift the equilibrium. |
| **The Stress** | **[CO2]** | **Right or Left** | **Reactants or Products** |
| CaCO3 is added |  |  |  |
| CaO is added |  |  |  |
| CO2 is added |  |  |  |
| Temp is decreased |  |  |  |
| A catalyst is added |  |  |  |
| [CO2] is decreased |  |  |  |
| Temp is increased |  |  |  |
| CaO is removed |  |  |  |