

Target: I can pay attention to small details when doing Le Chatelier's principle problems

Things to look for BEFORE answering an equilibrium problem		
Stressor	Question	What does it tell us?
Increase or decrease [] products or reactants	Which phase?	<ul style="list-style-type: none">• Gas, aqueous - change things• Solid, Liquid – DON'T CHANGE ANYTHING!
Increase or decrease T	Endo or exo?	<ul style="list-style-type: none">• Endo = absorbed, so it is a REACTANT• Exo = released, so it is a PRODUCT
Increase or decrease total Pressure <i>(Same as change in volume or number of moles of gas)</i>	How many moles of GAS are on each side of the equation?	<ul style="list-style-type: none">• Increase pressure = move to side with FEWER moles of gas• Decrease pressure = move to side with MORE moles of gas