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| **Equilibrium** | **Le Chatelier’s Principle** |
| **Too cold?**  **N2 (g) + 3H2 (g) 🡪 2NH3 (g) + 92.05 KJ** | **Too hot?**  **N2 (g) + 3H2 (g) 🡪 2NH3 (g) + 92.05 KJ** |
| **Too reactants?**  **N2 (g) + 3H2 (g) 🡪 2NH3 (g) + 92.05 KJ** | **Too many products?**  **N2 (g) + 3H2 (g) 🡪 2NH3 (g) + 92.05 KJ** |
| **Too much pressure?**  **N2 (g) + 3H2 (g) 🡪 2NH3 (g) + 92.05 KJ**  *\*only care about\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_* | **Too little pressure?**  **N2 (g) + 3H2 (g) 🡪 2NH3 (g) + 92.05 KJ**  *\*only care about\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_* |
| **Equilibrium Hashtag:**  # | **Equilibrium Knock-Knock Joke:**  **Person 1:** Knock Knock!  **Person 2:** Who’s there?  **Person 1: \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ Person 2: \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_** who**?**  **Person 3: \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_** |