

Keys to Practice p. 64-67

# Get out your GREEN pen please!

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1) #-ing is off a bit,  
oh well! ☺  
2) Loses pieces to  
become more  
stable  
3) From notes!

4) Gamma 12) 210/81  
5) Alpha Tl, alpha  
6) Beta 13) 212/84  
7) Beta Po, beta

8) Gamma

9) 4/2 He,  
alpha

10) 0/-1 e,  
beta

11) 4/2 He,  
alpha

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1) 4/2 He,  
alpha

2) 0/-1 e,  
beta

3) 230/89  
Ac

4) 216/84  
Po

5) 216/85  
At

6) 0/-1 e,  
210/84

Po

7) Np turns  
into Pa

but still  
isn't  
stable.

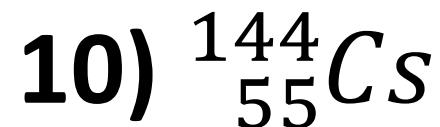
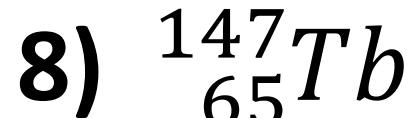
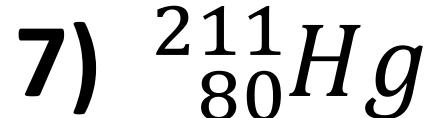
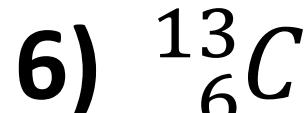
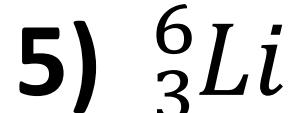
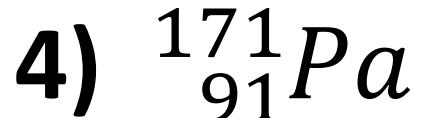
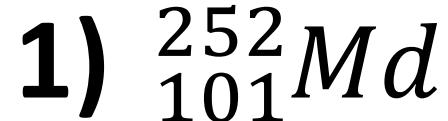
Keeps  
under

going  
decays  
until it is

more  
stable. It  
is a  
“chain  
decay”  
or a  
“series  
decay”  
or a  
“decay  
series”

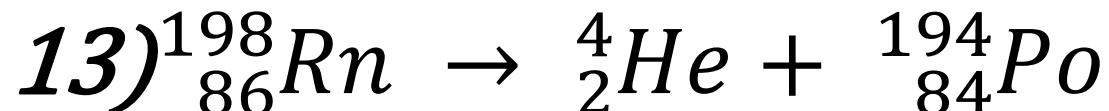
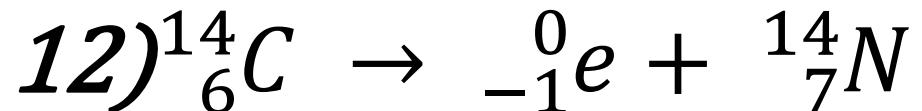
# Get out your GREEN pen please!

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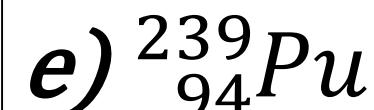
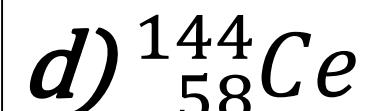
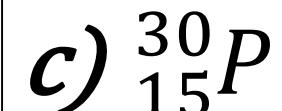
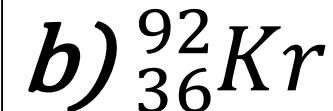


# Get out your GREEN pen please!

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- 1) Two. Proton & neutron
- 2) because electrons are so small we dont count them
- 3) skip
- 4)
  - a)  ${}_{5}^{11}B$
  - b)  ${}_{2}^{4}He$
  - c)  ${}_{88}^{234}Ra$
  - d)  ${}_{-1}^0e$
  - e)  ${}_{56}^{141}Ba$
  - f)  ${}_{90}^{234}Th$
  - g)  ${}_{58}^{146}Ce$
  - h)  ${}_{2}^{4}He$

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