Neutral Compound Practice

To write a neutral compound, all charges must \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Ionic compounds are written with the \_\_\_\_\_\_\_\_\_\_\_\_\_ first, and the \_\_\_\_\_\_\_\_\_\_\_\_ second.

|  |  |  |  |  |  |  |
| --- | --- | --- | --- | --- | --- | --- |
|  | **Iodide** | **Fluoride** | **Oxide** | **Phosphide** | **Nitrate** | **Sulfate** |
| **Calcium** |  |  |  |  |  |  |
| **Silver** |  |  |  |  |  |  |
| **Barium** |  |  |  |  |  |  |
| **Lead (II)** |  |  |  |  |  |  |
| **Chromium(III)** |  |  |  |  |  |  |
| **Tin (IV)** |  |  |  |  |  |  |
| **Ammonium** |  |  |  |  |  |  |
| **Copper (II)** |  |  |  |  |  |  |
| **Iron (II)** |  |  |  |  |  |  |
| **Aluminum** |  |  |  |  |  |  |
| **Magnesium** |  |  |  |  |  |  |
| **Lithium** |  |  |  |  |  |  |
| **Iron (III)** |  |  |  |  |  |  |
| **Vanadium(V)** |  |  |  |  |  |  |