Rules for Balancing Equations
1) Get a pencil! Stay calm.

2) Write the skeleton equation – this

 means convert the names of chemicals

 into formulas.

3) Be sure that all formulas are written

 correctly.

4) Count atoms of each element on each

 side of the arrow to figure out what

 needs to be balanced.

5) Change ONLY coefficients to balance,

 ***NEVER change subscripts!***

6) Reduce coefficients to lowest ratio.

 (i.e. 2:4 ration, is 1:2)

7) Check your work when done.

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