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| Table 1 reaction 4  CH4 + O2 → |  | → CO2 + H2O |
| Table 1 reaction 1  Na + O2 → |  | → Na2O |
| Table 1 reaction 3  Ag2O → |  | → Ag + O2 |
| Table 1 reaction 2  Al + Pb(NO3)2 → |  | → Pb + Al(NO3)3 |
| Table 1 reaction 5  K2CO3 + BaCl2 → |  | → KCl + BaCO3 |
| Table 2 reaction 3  C2H6 + O2 → |  | → CO2 + H2O |
| Table 2 reaction 2  Ra + Cl2 → |  | → RaCl2 |
| Table 2 reaction 5  Al(OH)3 → |  | → Al2O3 + H2O |
| Table 2 reaction 1  Cl2 + NaI → |  | → NaCl + I2 |
| Table 2 reaction 4  Al2(SO4)3 + Ca3(PO4)2 → |  | → AlPO4 + CaSO4 |
| Table 3 reaction 3  P2O3 + H2O → |  | → H3PO3 |
| Table 3 reaction 2  C3H8 + O2 → |  | → CO2 + H2O |
| Table 3 reaction 4  MgO → |  | → Mg + O2 |
| Table 3 reaction 1  Mg + HCl → |  | → MgCl2 + H2 |
| Table 3 reaction 5  Al(OH)3 + HC2H3O2 → |  | → Al(C2H3O2)3 + H2O |
| Table 4 reaction 1  C4H10 + O2 → |  | → CO2 + H2O |
| Table 4 reaction 4  BaO + CO2 → |  | → BaCO3 |
| Table 4 reaction 3  H2SO4 → |  | → SO3 + H2O |
| Table 4 reaction 2  Fe + CuSO4 → |  | → FeSO4 + Cu |
| Table 4 reaction 5  Ca(OH)2 + H3PO4 → |  | → Ca3(PO4)2 + H2O |
| Table 5 reaction 1  Sodium + Copper (II) Oxide 🡪 |  | Table 5 reaction 2  Glucose + oxygen 🡪  C6H12O6 |
| Table 5 reaction 3  Calcium Chloride 🡪 |  | Table 5 reaction 4  aluminum chloride + Calcium Oxide 🡪 |
| Table 5 reaction 5  aluminum + Oxygen 🡪 |  | Table 5 reaction 1  Sodium + Copper (II) Oxide 🡪 Sodium oxide + Copper  2Na + CuO 🡪 Na2O + Cu  Single Dis |
| Table 5 reaction 2  glucose + oxygen 🡪 carbon dioxide + dihydrogen monoxide  C6H12O6 + 6O2 🡪 6CO2 + 6H2O  Combustion |  | Table 5 reaction 3  Calcium Chloride 🡪 Calcium + Chlorine  CaCl2 🡪 Ca + Cl2  Decomposition |
| Table 5 reaction 4  aluminum chloride + calcium oxide 🡪 aluminum oxide + calcium chloride  2AlCl3 + 3CaO 🡪 Al2O3 + 3CaCl2  Double Dis |  | Table 5 reaction 5  aluminum + Oxygen 🡪 aluminum oxide  4Al + 3O2 🡪 2Al2O3  Synthesis |