Guided Stoichiometry Practice

Mole Ratios:

__N_2 + ___H_2 → ___NH_3

60 mol H_2 → mol N_2

__N_2 + ___H_2 → ___NH_3

54 g N_2 → g NH_3

__N_2 + ___H_2 → ___NH_3

6.05 x 10^{57} molec NH_3 → molec N_2

__N_2 + ___H_2 → ___NH_3

4.12 x 10^{12} molec H_2 → g NH_3

__N_2 + ___H_2 → ___NH_3

Link to YouTube Presentation: https://youtu.be/nYS8hRWMRCQ