|  |
| --- |
| Stoichiometry Practice **\_\_N2 + \_\_\_H2 🡪 \_\_\_NH3** |
| **1** | 30 mol H2 🡪 mol NH3 |
| **2** | 15 mol N2 🡪 mol H2 |
| **3** | 82 g H2 🡪 g NH3 |
| **4** | 16 g NH3 🡪 g H2 |
| **5** | 5.35 x 1013 molec H2 🡪 molec NH3 |
| **6** | 3.25 x 1059 molec NH3 🡪 molec H2  |
| **7** | 2.45 x 1011 molec H2 🡪 g H2 |

|  |
| --- |
| Stoichiometry Practice **\_\_N2 + \_\_\_H2 🡪 \_\_\_NH3** |
| **1** | 30 mol H2 🡪 mol NH3 |
| **2** | 15 mol N2 🡪 mol H2 |
| **3** | 82 g H2 🡪 g NH3 |
| **4** | 16 g NH3 🡪 g H2 |
| **5** | 5.35 x 1013 molec H2 🡪 molec NH3 |
| **6** | 3.25 x 1059 molec NH3 🡪 molec H2  |
| **7** | 2.45 x 1011 molec H2 🡪 g H2 |

|  |
| --- |
| Stoichiometry Practice **\_\_N2 + \_\_\_H2 🡪 \_\_\_NH3** |
| **1** | 30 mol H2 🡪 mol NH3 |
| **2** | 15 mol N2 🡪 mol H2 |
| **3** | 82 g H2 🡪 g NH3 |
| **4** | 16 g NH3 🡪 g H2 |
| **5** | 5.35 x 1013 molec H2 🡪 molec NH3 |
| **6** | 3.25 x 1059 molec NH3 🡪 molec H2  |
| **7** | 2.45 x 1011 molec H2 🡪 g H2 |

|  |
| --- |
| Stoichiometry Practice **\_\_N2 + \_\_\_H2 🡪 \_\_\_NH3** |
| **1** | 30 mol H2 🡪 mol NH3 |
| **2** | 15 mol N2 🡪 mol H2 |
| **3** | 82 g H2 🡪 g NH3 |
| **4** | 16 g NH3 🡪 g H2 |
| **5** | 5.35 x 1013 molec H2 🡪 molec NH3 |
| **6** | 3.25 x 1059 molec NH3 🡪 molec H2  |
| **7** | 2.45 x 1011 molec H2 🡪 g H2 |

|  |
| --- |
| Stoichiometry Practice **\_\_N2 + \_\_\_H2 🡪 \_\_\_NH3** |
| **1** | 30 mol H2 🡪 mol NH3 |
| **2** | 15 mol N2 🡪 mol H2 |
| **3** | 82 g H2 🡪 g NH3 |
| **4** | 16 g NH3 🡪 g H2 |
| **5** | 5.35 x 1013 molec H2 🡪 molec NH3 |
| **6** | 3.25 x 1059 molec NH3 🡪 molec H2  |
| **7** | 2.45 x 1011 molec H2 🡪 g H2 |

|  |
| --- |
| Stoichiometry Practice **\_\_N2 + \_\_\_H2 🡪 \_\_\_NH3** |
| **1** | 30 mol H2 🡪 mol NH3 |
| **2** | 15 mol N2 🡪 mol H2 |
| **3** | 82 g H2 🡪 g NH3 |
| **4** | 16 g NH3 🡪 g H2 |
| **5** | 5.35 x 1013 molec H2 🡪 molec NH3 |
| **6** | 3.25 x 1059 molec NH3 🡪 molec H2  |
| **7** | 2.45 x 1011 molec H2 🡪 g H2 |

|  |
| --- |
| Stoichiometry Practice **\_\_N2 + \_\_\_H2 🡪 \_\_\_NH3** |
| **1** | 30 mol H2 🡪 mol NH3 |
| **2** | 15 mol N2 🡪 mol H2 |
| **3** | 82 g H2 🡪 g NH3 |
| **4** | 16 g NH3 🡪 g H2 |
| **5** | 5.35 x 1013 molec H2 🡪 molec NH3 |
| **6** | 3.25 x 1059 molec NH3 🡪 molec H2  |
| **7** | 2.45 x 1011 molec H2 🡪 g H2 |

|  |
| --- |
| Stoichiometry Practice **\_\_N2 + \_\_\_H2 🡪 \_\_\_NH3** |
| **1** | 30 mol H2 🡪 mol NH3 |
| **2** | 15 mol N2 🡪 mol H2 |
| **3** | 82 g H2 🡪 g NH3 |
| **4** | 16 g NH3 🡪 g H2 |
| **5** | 5.35 x 1013 molec H2 🡪 molec NH3 |
| **6** | 3.25 x 1059 molec NH3 🡪 molec H2  |
| **7** | 2.45 x 1011 molec H2 🡪 g H2 |