Stoichiometry Practice

__N_2 + ___H_2 → ___NH_3

1	30 mol H_2 → mol NH_3
2	15 mol N_2 → mol H_2
3	82 g H_2 → g NH_3
4	16 g NH_3 → g H_2
5	5.35×10^{13} molec H_2 → molec NH_3
6	3.25×10^{59} molec NH_3 → molec H_2
7	2.45×10^{11} molec H_2 → g H_2

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