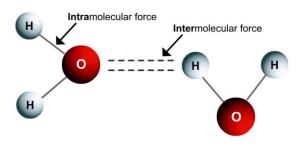
## 13 • IMFs, Liquids and Solids

## SAMPLE QUIZ QUESTIONS

1. Give names to the two forces mentioned in the tutorials and indicated in the picture on the right.

Intramolecular force:

Intermolecular force: \_\_\_\_\_



2. Put these three ionic compounds in order of **weakest** to **strongest** IMF.

\_\_\_\_\_



3. Define "miscible".

Are ammonia (NH<sub>3</sub>) and water (H<sub>2</sub>O) miscible? \_\_\_\_\_ Justify your answer.

- 4. Consider the **phase diagram** shown:
  - a) Label the x & y axes.
  - b) Is this water or some other substance?
  - c) Label the areas as gas, liquid, and solid.

