Name:	Data	Dominde	Soot #.
Name.	Date.	I ci iou.	Seat #.

Ornament Chemistry: Write the balanced chemical equation for the following. Indicate the oxidation numbers of the elements.

$$Zn(s) + HCl(aq) = \underline{\hspace{1cm}} + \underline{\hspace{1cm}}$$

	Reactants	Products
Zn		
Н		
Cl		

$$Fe(s) + CuSO_4(aq) = ___ + ____$$

	Reactants	Products
Fe		
Cu		
S		
О		

☐ Terminology:	:
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In the first reaction, ______ is getting oxidized. ______ is the oxidizing agent.

In the first reaction, ______ is getting reduced. ______ is the reducing agent.

In the second reaction, ______ is getting oxidized. ______ is the oxidizing agent.

In the second reaction, ______ is getting reduced. ______ is the reducing agent.