MOLARITY EXTRA PRACTICE WORKSHEET			
1. What is the molarity of 2.0 L of solution made from 2.4 moles of Nat 2. What is the molarity of 500. mL of solution made from 0.70 moles of 3. What is the molarity of 100. mL of solution made from 0.45 g of Nat	CI and water? f LiCI and water? DH and water?		
4. How many moles of NaCl are found in 2.2 L of a 0.300 M solution?	ion?		
6. What volume of a 1.40 M solution of H_2SO_4 contains 0.400 moles of	f H₂SO₄?		
7. How many moles of sugar are needed to make 60. mL of a 0.25 M	solution?		
8. What mass of CH ₃ OH is needed to prepare 1.50 L of a 3.00 M solution 9. What mass of K_3PO_4 is needed to make 4.00 L of a 1.50 M solution	ition? 1?		
10. What volume of a 0.750 M solution can be made from 90.0 g of N	H₄CI?		
Answers: 1) 1.2 M 2) 1.4 M 3) 0.11 M 4) 0.66 mol 5) 0.007	5 mol 6) 0.286 L	7) 0.015 mol 8) 144 g	9) 1270
MOLARITY EXTRA PRACTICE WORKSHEET			
1. What is the molarity of 2.0 L of solution made from 2.4 moles of Na	Cl and water?		
3. What is the molarity of 100. mL of solution made from 0.45 g of Nat	OH and water?		
4. How many moles of NaCl are found in 2.2 L of a 0.300 M solution?	ion2		
6. What volume of a 1.40 M solution of H_2SO_4 contains 0.400 moles of	f H₂SO₄?		
7. How many moles of sugar are needed to make 60. mL of a 0.25 M	solution?		
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10. What volume of a 0.750 M solution can be made from 90.0 g of N	l₄CI?		
Answers: 1) 1.2 M 2) 1.4 M 3) 0.11 M 4) 0.66 mol 5) 0.007	5 mol 6) 0.286 L	7) 0.015 mol 8) 144 g	9) 1270
MOLARITY EXTRA PRACTICE WORKSHEET			
1. What is the molarity of 2.0 L of solution made from 2.4 moles of Na	Cl and water?		
3. What is the molarity of 100. mL of solution made from 0.45 g of Nat	OH and water?		
4. How many moles of NaCl are found in 2.2 L of a 0.300 M solution?			
5. How many moles of C_2H_5OH are found in 50. mL of a 0.150 M solution 6. What volume of a 1.40 M solution of H_2SO_4 contains 0.400 moles of	ion <i>?</i> f H₂SO₄?		
7. How many moles of sugar are needed to make 60. mL of a 0.25 M	solution?		
8. What mass of CH_3OH is needed to prepare 1.50 L of a 3.00 M solution 9. What mass of K_2PO_4 is needed to make 4.00 L of a 1.50 M solution	ition? 12		
10. What volume of a 0.750 M solution can be made from 90.0 g of NI	⊣₄CI?		
Answers: 1) 1.2 M 2) 1.4 M 3) 0.11 M 4) 0.66 mol 5) 0.007	5 mol 6) 0.286 L	7) 0.015 mol 8) 144 g	9) 1270
MOLARITY EXTRA PRACTICE WORKSHEET			
1. What is the molarity of 2.0 L of solution made from 2.4 moles of National 2. What is the molarity of 500 ml of solution mode from 0.70 moles a	Cl and water?		
3. What is the molarity of 100. mL of solution made from 0.45 g of Nat	OH and water?		
4. How many moles of NaCl are found in 2.2 L of a 0.300 M solution?			
5. How many moles of C_2H_5OH are found in 50. mL of a 0.150 M solution of H ₂ SO ₄ contains 0.400 moles of	ion? f H₂SO₄?		
7. How many moles of sugar are needed to make 60. mL of a 0.25 M	solution?		
8. What mass of CH_3OH is needed to prepare 1.50 L of a 3.00 M solution 9. What mass of K_2PO_4 is needed to make 4.00 L of a 1.50 M solution	ition? 12		
10. What volume of a 0.750 M solution can be made from 90.0 g of NI	H₄CI?		
Answers: 1) 1.2 M 2) 1.4 M 3) 0.11 M 4) 0.66 mol 5) 0.007	5 mol 6) 0.286 L	7) 0.015 mol 8) 144 g	9) 1270
MOLARITY EXTRA PRACTICE WORKSHEET			
1. What is the molarity of 2.0 L of solution made from 2.4 moles of Na	Cl and water?		
3. What is the molarity of 100. mL of solution made from 0.45 g of Nat	OH and water?		
4. How many moles of NaCl are found in 2.2 L of a 0.300 M solution?			
6. What volume of a 1.40 M solution of H ₂ SO ₄ contains 0.400 moles of H ₂ SO ₄ ?			
7. How many moles of sugar are needed to make 60. mL of a 0.25 M solution?			
8. what mass of CH_3OH is needed to prepare 1.50 L of a 3.00 M solution 9. What mass of K ₂ PO ₄ is needed to make 4.00 L of a 1.50 M solution	ונוסח? ו?		
10. What volume of a 0.750 M solution can be made from 90.0 g of N	l₄Cl?		
Answers: 1) 1.2 M 2) 1.4 M 3) 0.11 M 4) 0.66 mol 5) 0.007	5 mol 6) 0.286 L	7) 0.015 mol 8) 144 g	9) 1270