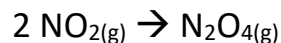


ICE Table Practice Problem #2

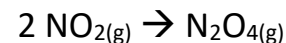
In the following reaction, $K_{eq} = 9.3 \times 10^{-7}$ at room temp. Calculate the equilibrium concentration of N_2O_4 in a flask initially containing only 3.00 M of NO_2



Rxn	$2 NO_{2(g)} \rightarrow N_2O_{4(g)}$	
I		
C		
E		
5%		
Answer		

ICE Table Practice Problem #2

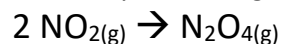
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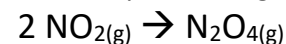
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