Things to look for BEFORE answering an equilibrium $ {f Q} $		
Stressor	Question	What does it tell us?
Increase or decrease [] products or reactants	Which phase?	 Gas, aqueous - change things Solid, Liquid – DON'T CHANGE ANYTHING!
Increase or decrease T	Endo or exo?	 Endo = absorbed, so it is a REACTANT Exo = released, so it is a PRODUCT
Increase or decrease total Pressure	How many moles of GAS are on each side of the equation?	 Increase pressure = move to side with FEWER moles of gas
(Same as ∆ in V or # of moles of gas)		 Decrease pressure = move to side with MORE moles of gas

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