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| **Weak Acid/Base Practice Problem #2**You have 0.010 M NH3. Calculate the pH. Kb = 1.8 x 10-5NH3 + H2O 🡨🡪 NH4+ + OH-

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| --- | --- |
| **Rxn** | NH3 🡨🡪 NH4+ + OH- |
| **I** |  |  |  |
| **C** |  |  |  |
| **E** |  |  |  |
| **5%** |  |  |  |
| **Answer** |  |  |  |

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