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| **Weak Acid/Base Practice Problem #1**You have 1.00 M HOAc. Calc. the equilibrium concentrations of HOAc, H3O+, OAc-, and the pH if Ka = 1.8x10-5.HOAc + H2O 🡨🡪 H3O + OAc-

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| --- | --- |
| **Rxn** | HOAc 🡨🡪 H3O + OAc- |
| **I** |  |  |  |
| **C** |  |  |  |
| **E** |  |  |  |
| **5%** |  |  |  |
| **Answer** |  |  |  |

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