Worksheet #10

Name:	Period: Seat#:
The questions are in chronological order.Times are estimates. You do not need full sentences.	
1) Where does an element take its identity from? (4:30)	8) How does the atomic arrangement of atoms lead to its crystal structure like was seen in the sample of bronze with gold (Au) and tin (Sn) atoms? (31:00)
2)How much gold (Au) is extracted per ton of rock ore? (7:30)	9) What is the atomic number and what does the atomic number indicate? (33:00)
3) How much does a gold (Au) bar weigh and how much is it worth? (12:00)	10) Most of the periodic table is made of what type of elements? (34:00)
4) Why is copper (Cu) so widely sought on the world market and New York Mercantile Exchange? (15:00)	11) How did early chemists like Mendeleev classify the elements? (37:00)
5) What is copper (Cu) combined with to make bronze? (17:00)	12) How is the periodic table structured with regard to elements with similar properties? (39:00)
6) What makes metals like Copper (Cu) conductive to electricity? (19:00)	13) What makes noble gases stable? (42:00)
7) Bronze is an alloy. What is an alloy and why are they	14) Why is an alkali metal element like Sodium (Na) so

- **7)**Bronze is an alloy. What is an alloy and why are they preferable at times? (21:00)
- **14)** Why is an alkali metal element like Sodium (Na) so reactive? (44:00)

- **15)** What does chlorine (Cl⁻) do for sodium (Na⁺)? What 22) How does silicon shape our technological reality? tasty substance is produced when this happens? (47:00) (1:21:00)16) What powers explosions and fire? (54:00) 23) How are rare earth elements like neodymium (Nd) important to our technological world? (1:26:00) 17) What elements are basic to all living things? (58:00) 24) What is an isotope like Carbon-14? (1:41:00) 25) How can an isotope like Carbon-14 be used to date **18)** Why is Carbon (C) so good for forming the structure dead organisms? (1:43:00). of life? (1:05:00). **26)** What is an unstable radioactive isotope? (1:45:00) **19)** What are at least three (3) other elements that are used for life functions and what are their uses? (1:11:00) **20)** Why are cyanobacteria from places like volcanic 27) Why don't the man-made radioactive elements exist pools so important for the production of oxygen in our for very long? (1:57:00) atmosphere? (1:16:00)
- **21)** What was the original element formed moments after the Big Bang? What then created higher order elements? (1:18:00)