**Title, descriptive, not wordy. NOT a sentence. WHAT you did and HOW you did it.**

Your name – Contribution

Group Member Name – Contribution

Group Member Name – Contribution

Group Member Name – Contribution

*May 21, 2012*

*College Preparatory Chemistry, Sixth Period*

**Abstract**

*Paragraph form. Short, concise. What was the purpose of the lab? What were your results? How were your results determined? TECHNIQUES! Report accepted values/answer and percent error if applicable. Any conclusions?*  
The density of an irregular shaped metal block was found to be 1.2g/mL. The mass was found via digital scale, and the volume was found by using the technique of water displacement. The accepted value was 1.3g/mL, therefore there was a 7.69% error. Considering a relatively low percent error, finding the volume through water displacement was found to be a sufficiently accurate way to obtain the volume to use in density calculations.