

Title, descriptive, not wordy. NOT a sentence. WHAT you did and HOW you did it.

Your name – Contribution

Group Member Name – Contribution

Group Member Name – Contribution

Group Member Name – Contribution

May 21, 2012

College Preparatory Chemistry, Sixth Period

Abstract

Paragraph form. Short, concise. What was the purpose of the lab? What were your results? How were your results determined? TECHNIQUES! Report accepted values/answer and percent error if applicable. Any conclusions?

The density of an irregular shaped metal block was found to be 1.2g/mL. The mass was found via digital scale, and the volume was found by using the technique of water displacement. The accepted value was 1.3g/mL, therefore there was a 7.69% error. Considering a relatively low percent error, finding the volume through water displacement was found to be a sufficiently accurate way to obtain the volume to use in density calculations.